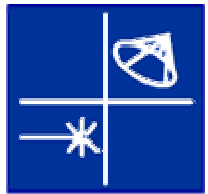


PIXE PAN 2007

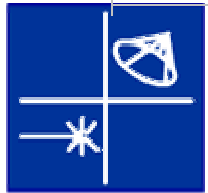
Introduction



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Schedule

	Mon	Tue	Wed	Thur	Fri
	6/18/2007	6/19/2007	6/20/2007	6/21/2007	6/22/2007
9:00	Intro (Mary)	PIXE INTRO	Lab 1/2 Intro	Lab 3/4 Intro	Lab 5 Intro
9:30	NSL Intro	PIXE	Lab 1/2	Lab 3 /4	Lab 5
10:00	and Tour				Lab Review/Makeup
11:00	Jordan Intro and Tour	Lunch	Lunch	Lunch	Lunch
12:00	Lunch at				
13:00	SDH	PIXE	Lab 2/1 Intro	Lab 4/3 Intro	Classroom Transfer
13:30	Rad Safety		Lab 2/1	Lab 4/3 Intro	(NSH 184)
14:00	Hands on Lab	Recap/Q&A	Recap/Q&A	Recap/Q&A	Introduction to Nuclear Physics
15:00	Introduction (JHS)				Review/ Plan
16:00	PIXE applications talk				for Student Week
16:30	NSL Conference Rm				
17:00	(Larry Lamm)				
	6/25/2007	6/26/2007	6/27/2007	6/28/2007	6/29/2007
9:00	Intro (Mary)	Lab 1 Intro	Lab 3 Intro	Lab 5 Intro	PIXE Applications (Lamm) (NSL)
9:30	NSL Intro	Lab 1	Lab 3	Lab 5	Analysis
10:00	and Tour				Presentation
11:00	Jordan Hall Intro/Tour	Lunch	Lunch	Lunch	Prep
12:00	Lunch at				Lunch
13:00	SDH	Lab2 Intro	Lab 4 Intro	Analysis	Student Presentations
13:30	Rad Safety	Lab 2	Lab 4		Presentation Preparation
14:00	Hands on Lab			Lab Make up	
14:30	Introduction				
15:00	(JHS)				
16:00	Introduction to JINA		Recap/Q&A		Digital Visualization Theater Presentation
16:30	Science (NSL Conf)		Recap/Q&A		Stellar Evolution and Supernovae
17:00	(Michael Wiescher)				Dr. Philip Sakimoto, Public Welcome



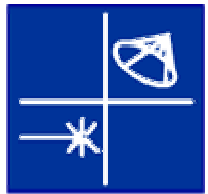
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General Comments

- Emphasis is not on getting the correct answer, but rather understanding the experimental methods used and physical concepts involved.
- All participants are encouraged to ask questions.

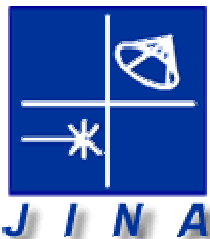
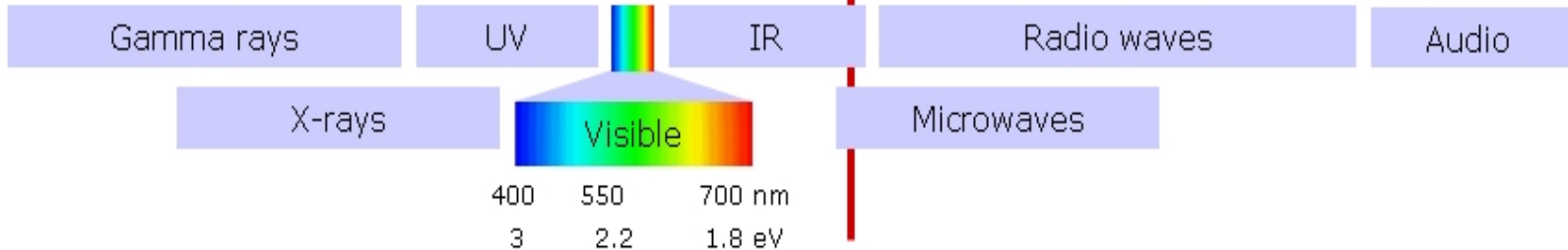


Photons

The Electromagnetic Spectrum

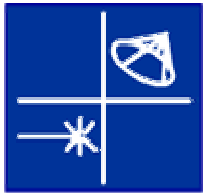
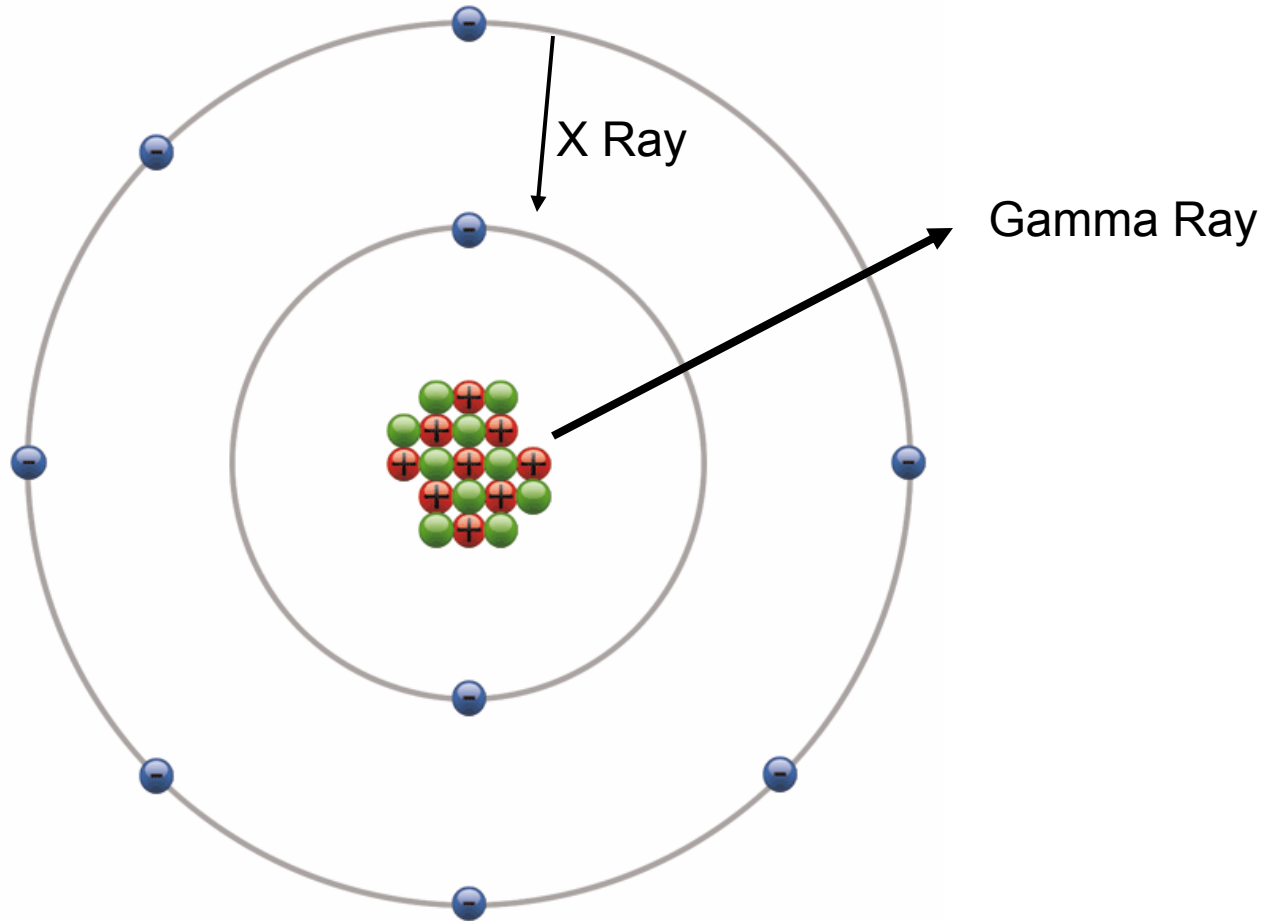
$k_B T_R$ -The thermal energy at room temperature

λ/m	10^{-13}	10^{-12}	10^{-11}	10^{-10}	10^{-9}	10^{-8}	10^{-7}	10^{-6}	10^{-5}	10^{-4}	10^{-3}	10^{-2}	10^{-1}	1	10^1	10^2	10^3	10^4	10^5	
	pm		Å	nm	μm			mm	m		km									
E/eV	10^7	10^6	10^5	10^4	10^3	10^2	10^1	1	10^{-1}	10^{-2}	10^{-3}	10^{-4}	10^{-5}	10^{-6}	10^{-7}	10^{-8}	10^{-9}			



* Illustration from Opensource Handbook of Nanoscience and Nanotechnology

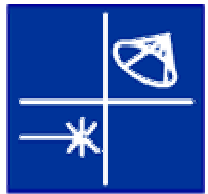
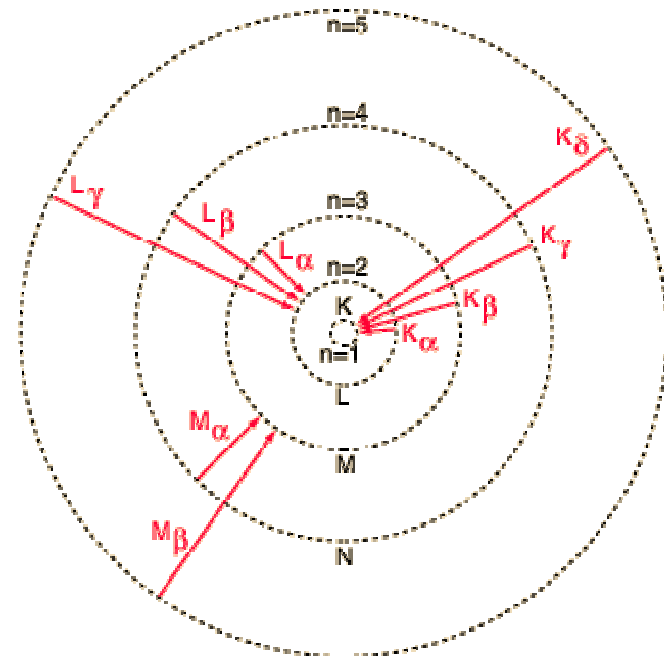
X and Gamma Ray Origins



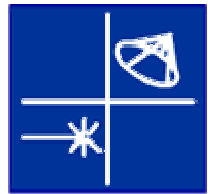
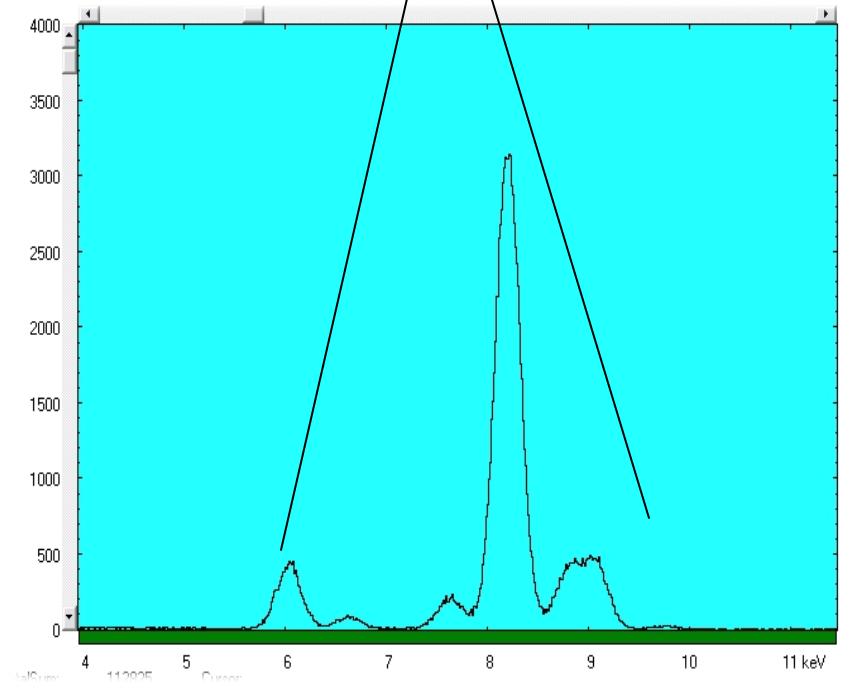
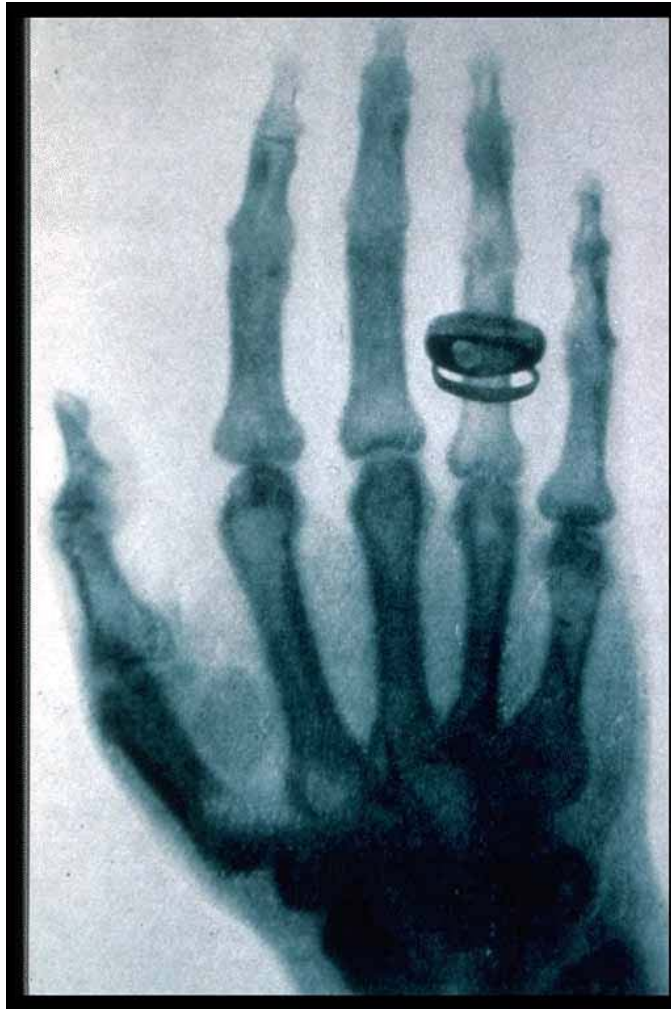
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Some X Ray Nomenclature

- The orbits (or shells) are most often referred to by a letter rather than the principle quantum number (n=1 is the K shell, n=2 is the L shell, etc.)
- Transitions are labeled K_{α} , K_{β} , etc. as shown.



Information from X-rays

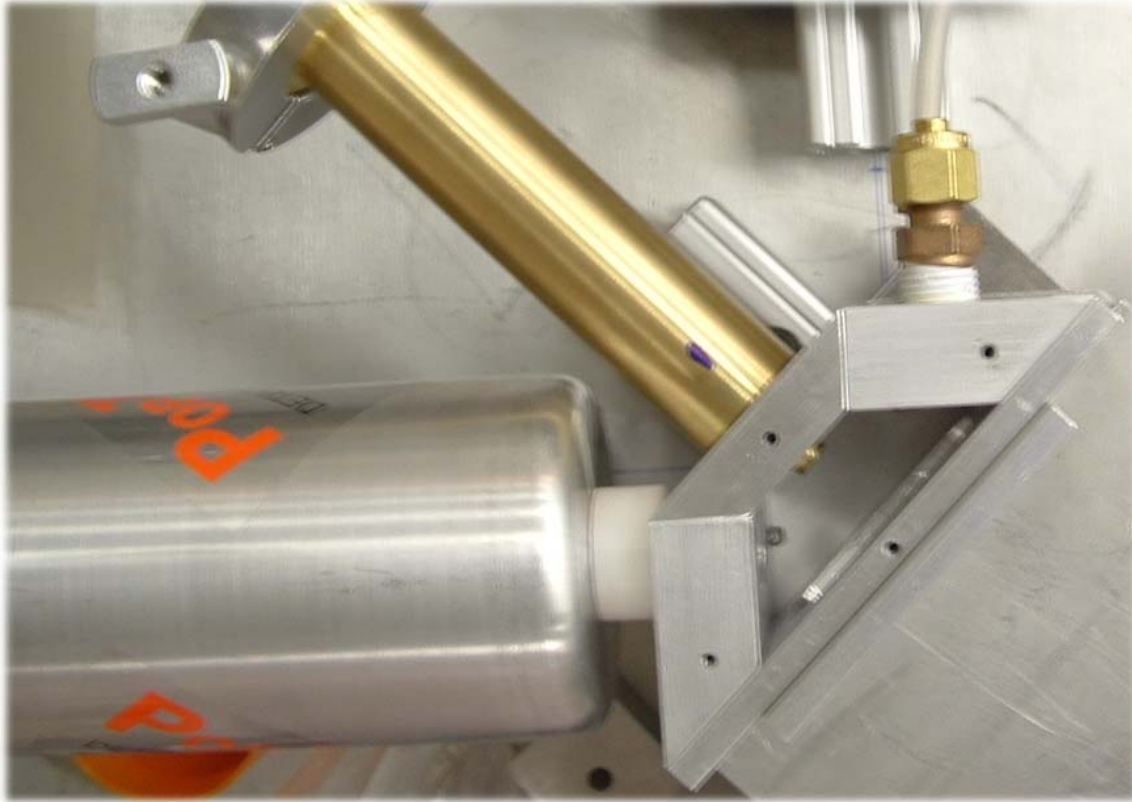


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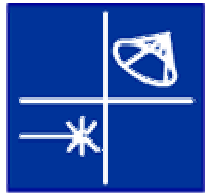
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Experiment I PIXE



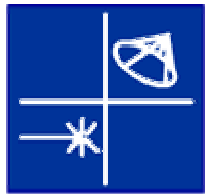
- Use a proton beam from the NSL's FN accelerator to excite a variety of targets
- Use the characteristic X Ray Energies to identify the elemental composition of the materials.
- Use the small beamspot (1-2mm diameter) to investigate spatial distributions.



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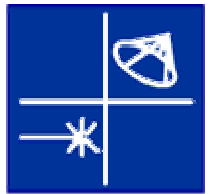
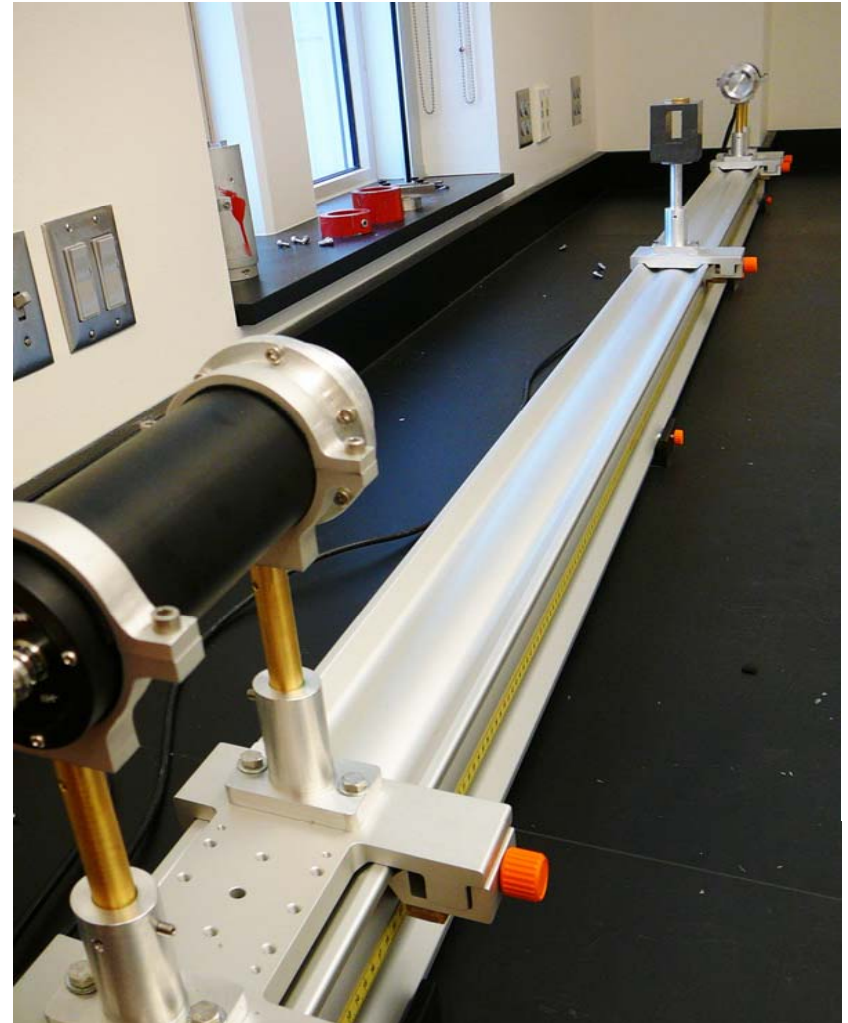
Experiment II XRF

- X Ray Fluorescence – Use an X-Ray source to excite atoms in the target instead of the proton beam.
- Measure a variety of objects.
- Test Moseley's Law relating X-ray Energy to atomic number.



Experiment III Speed of Light

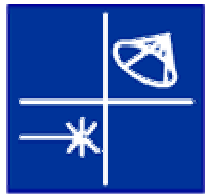
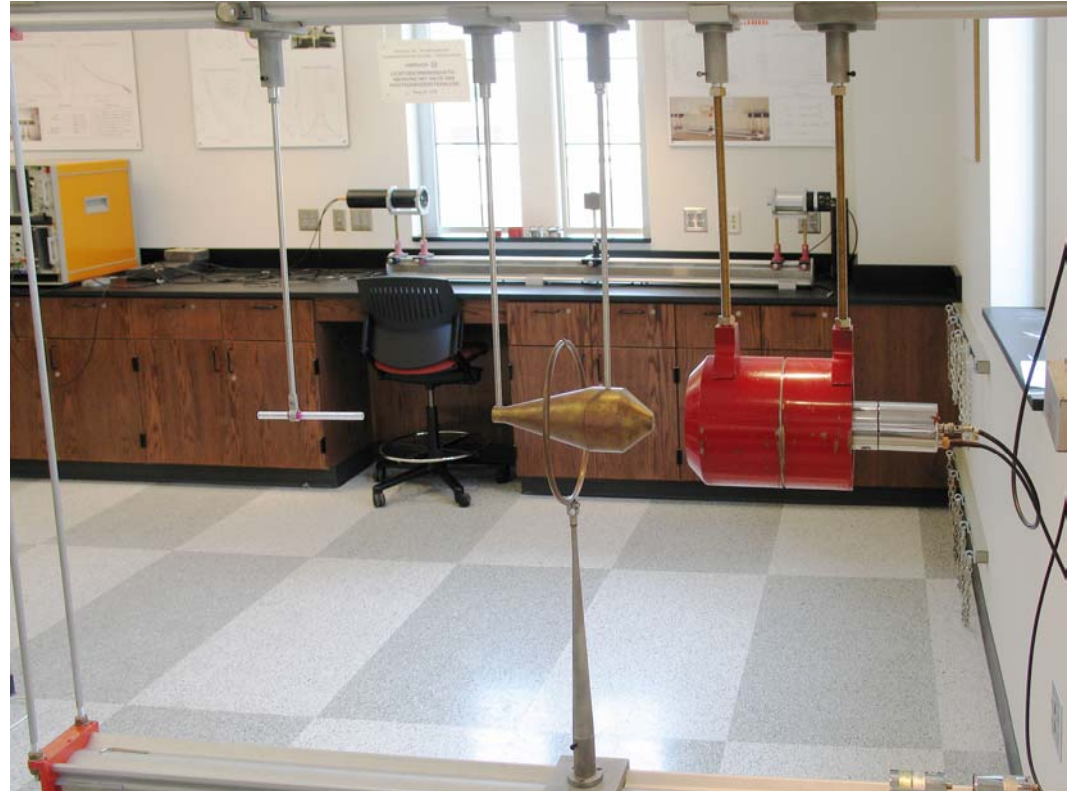
- Use Electronic modules as the stopwatch to time how long it takes a 511 keV photon to travel a distance of a meter or so.
- Use this information and a simple analysis to measure directly the speed of light.
- Photon source is the positron-electron annihilation from ^{22}Na .



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Experiment IV Compton Effect

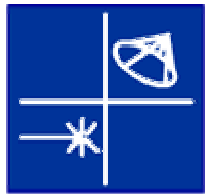
- An example of the particle nature of light by measuring the change in energy of photons scattered through different angles.



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Experiment V e/m measurement

- See the effect of magnetic and electric fields on a motion of an electron
- Determine the electrons charge to mass ratio



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