

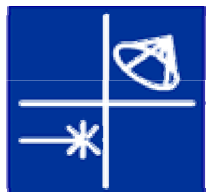
PIXE PAN 2008

Introduction



Schedule

Week 1: Teachers					
	Mon	Tue	Wed	Thur	Fri
	6/16/2008	6/17/2008	6/18/2008	6/19/2008	6/20/2008
9:00	Snacks/Welcome Intro (Mary DeWitt) JHS Atrium	PIXE INTRO	Lab 1/2 Intro	Lab 3/4 Intro	Lab 5 Intro
9:30	Jordan Intro and Tour	PIXE	Lab 1/2	Lab 3 /4	Lab 5
10:00	Rad Safety				Lab Review/Makeup
11:00	Rad Safety				
12:00	Lunch	Lunch	Lunch Seminar	Lunch	Lunch
13:00			Jay LaVerne	Lab 4/3 Intro	Introduction to Nuclear Physics Tony Hyder (NSH 184)
13:30	NSL Intro + Tour (Ed Stech)	PIXE	Lab 2/1 Intro	Lab 4/3 Intro	
14:00	PIXE applications talk		Lab 2/1		Review/ Plan for Student Week
15:00	NSL Conference Rm (Larry Lamm)	Recap/Q&A	Recap/Q&A	Recap/Q&A	
16:00	Recap/Q&A				
16:30					
17:00					
Week 2: Teachers and Students					
	6/23/2008	6/24/2008	6/25/2008	6/26/2008	6/27/2008
9:00	Snacks/Welcome Intro (Mary DeWitt) JHS Atrium	Lab 1 Intro	Lab 3 Intro	Lab 5 Intro	PIXE Applications (Lamm) (NSL)
9:30	Jordan Intro and Tour	Lab 1	Lab 3	Lab 5	Analysis Presentation Prep
10:00	Rad Safety				
11:00	Rad Safety				
12:00	Lunch	Lunch	Lunch	Lunch - NSH 202	Lunch
13:00		Lab2 Intro	Lab 4 Intro	(Astro: Peter Gamavich)	Student Presentations
13:30	NSL Intro + Tour (Ed Stech)	Lab 2	Lab 4	Analysis Presentation Preparation	184 NSH
14:00	Introduction to JINA Science (NSL Conf) (Philippe Collon)				Each group of students @ 15 minutes each
14:30	Recap/Q&A	Recap/Q&A	Recap/Q&A	Make Up Lab	Tour of the Labs
15:00					for parents by students.
16:00					
16:30					
17:00					



J I N A

General Comments

- Emphasis is not on getting the correct answer, but rather understanding the experimental methods used and physical concepts involved.
- All participants are encouraged to ask questions.
- Work as a group, help each other understand.

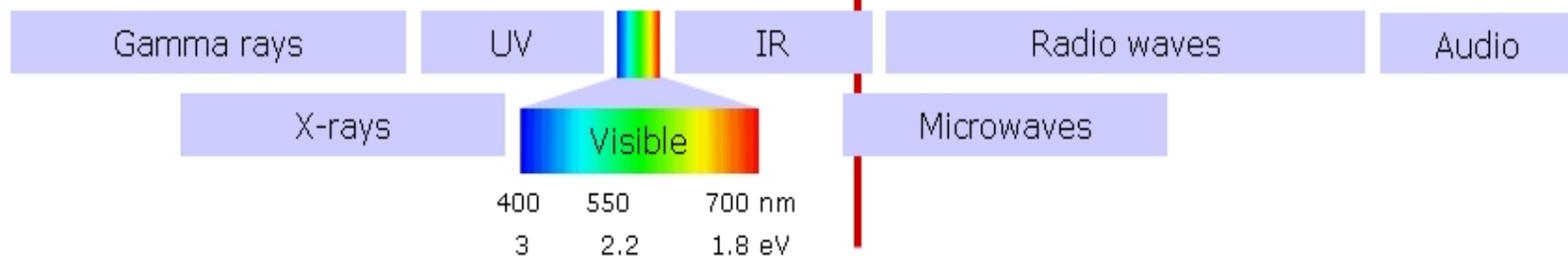


Photons

The Electromagnetic Spectrum

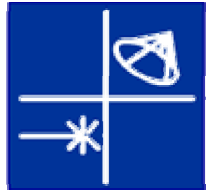
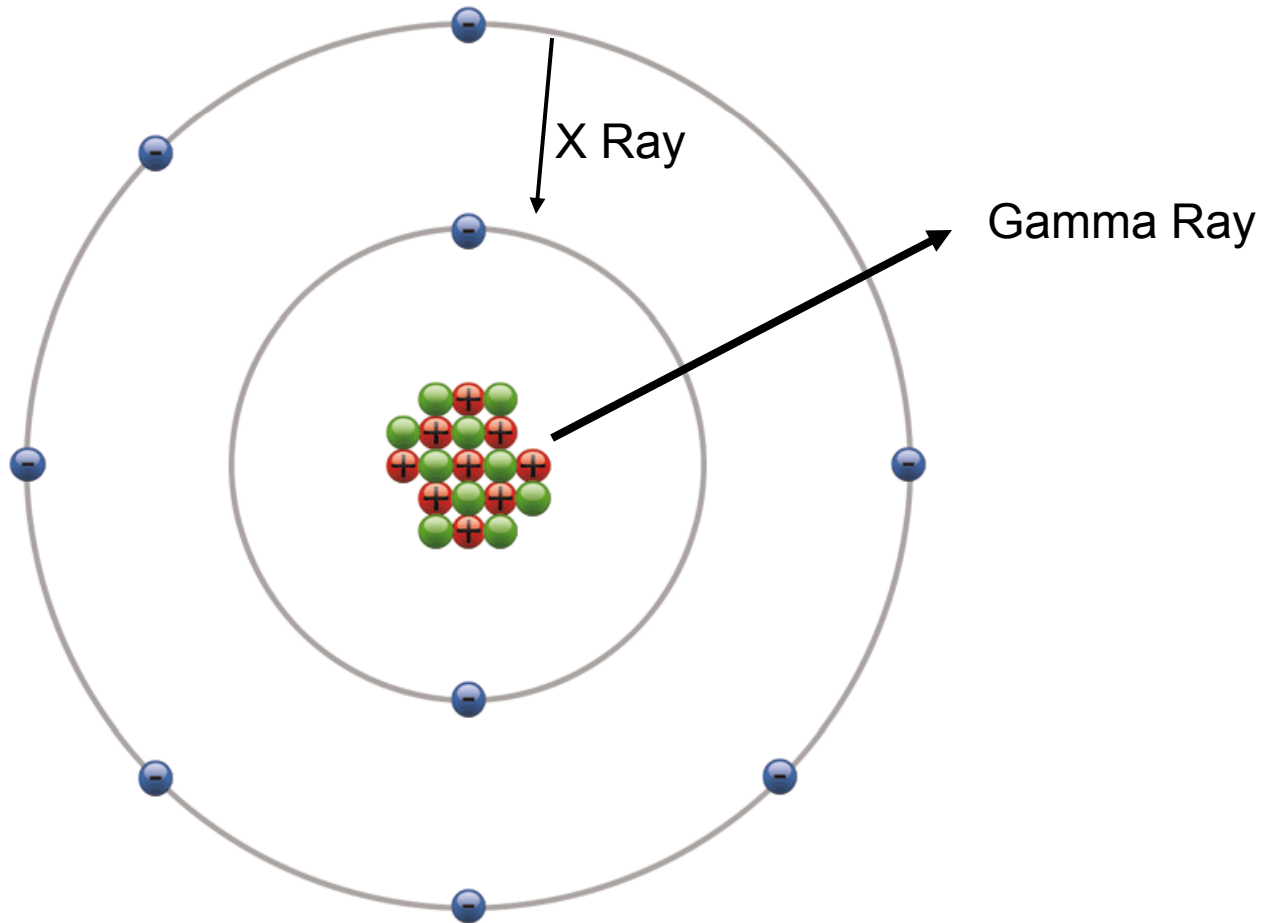
$k_B T_R$ -The thermal energy at room temperature

λ/m	10^{-13}	10^{-12}	10^{-11}	10^{-10}	10^{-9}	10^{-8}	10^{-7}	10^{-6}	10^{-5}	10^{-4}	10^{-3}	10^{-2}	10^{-1}	1	10^1	10^2	10^3	10^4	10^5
	pm		Å	nm		μm		mm		m		km							
E/eV	10^7	10^6	10^5	10^4	10^3	10^2	10^1	1	10^{-1}	10^{-2}	10^{-3}	10^{-4}	10^{-5}	10^{-6}	10^{-7}	10^{-8}	10^{-9}		



* Illustration from Opensource Handbook of Nanoscience and Nanoechnology

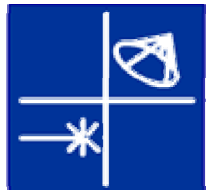
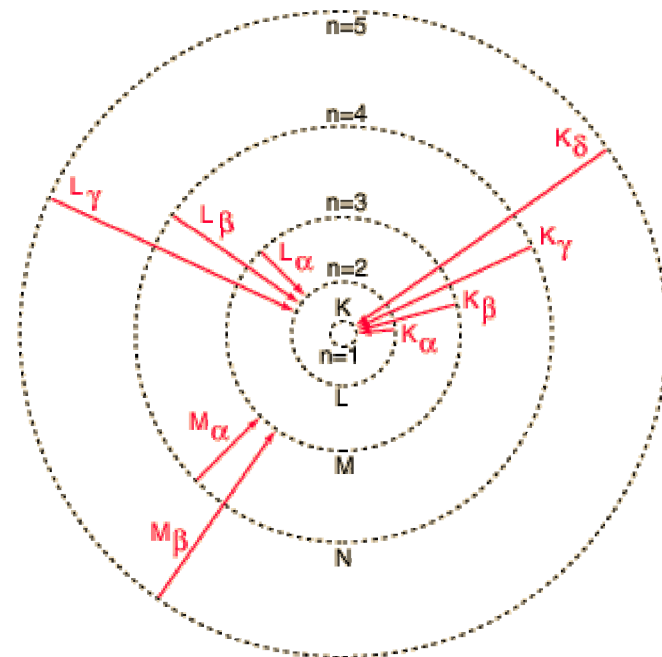
X and Gamma Ray Origins



J I N A

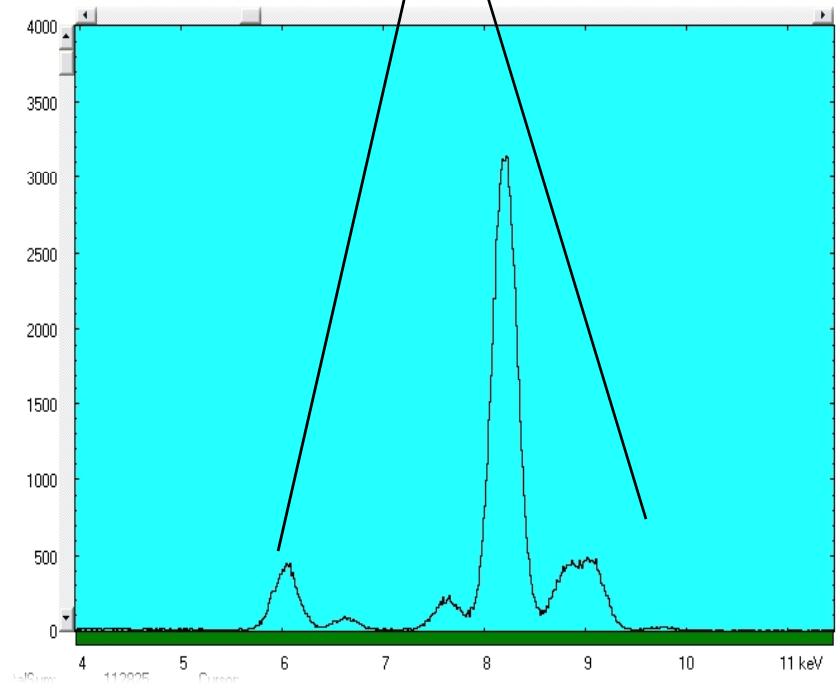
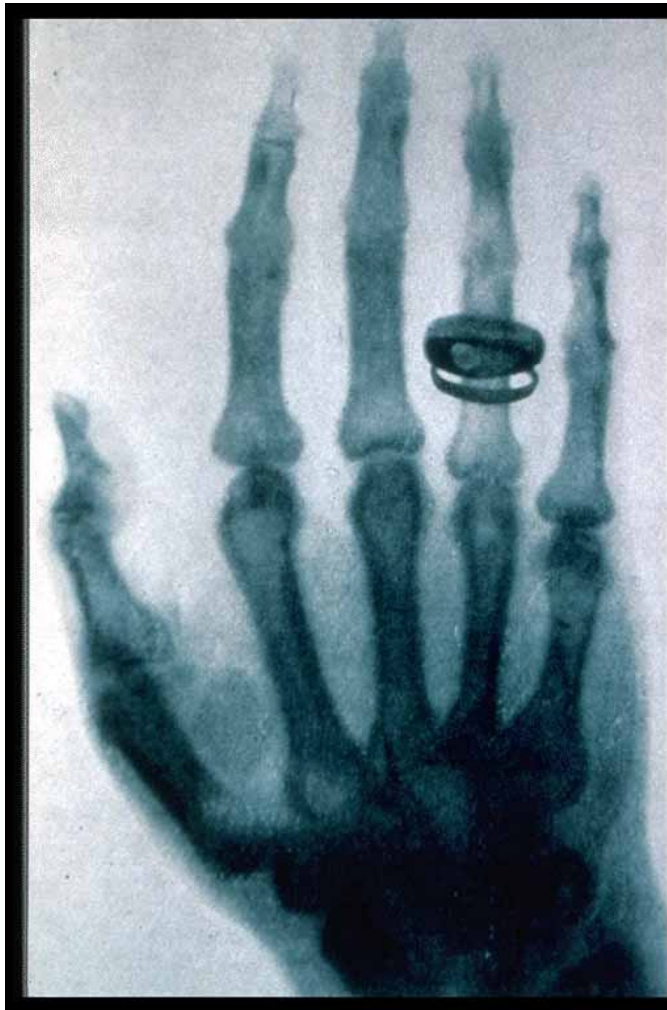
Some X Ray Nomenclature

- The orbits (or shells) are most often referred to by a letter rather than the principle quantum number (n=1 is the K shell, n=2 is the L shell, etc.)
- Transitions are labeled K_{α} , K_{β} , etc. as shown.



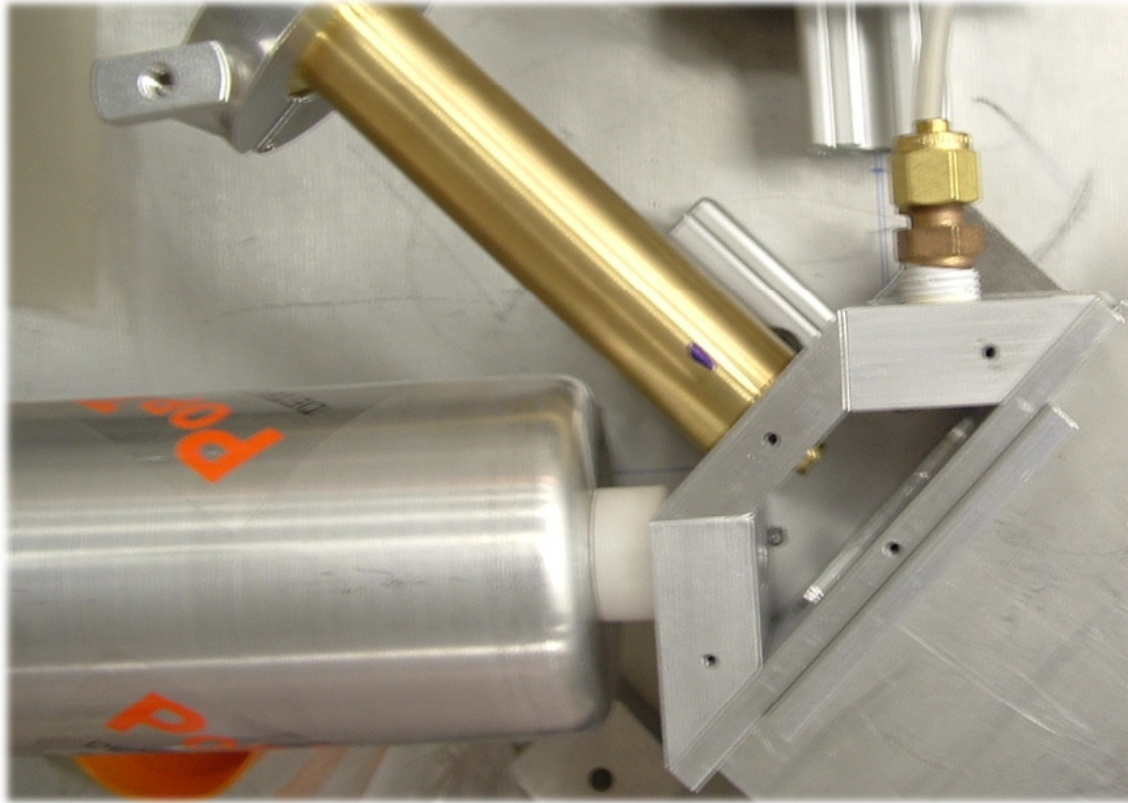
J I N A

Information from X-rays



J I N A

Experiment I PIXE



- Use a proton beam from the NSL's FN accelerator to excite a variety of targets
- Use the characteristic X Ray Energies to identify the elemental composition of the materials.
- Use the small beamspot (1-2mm diameter) to investigate spatial distributions.

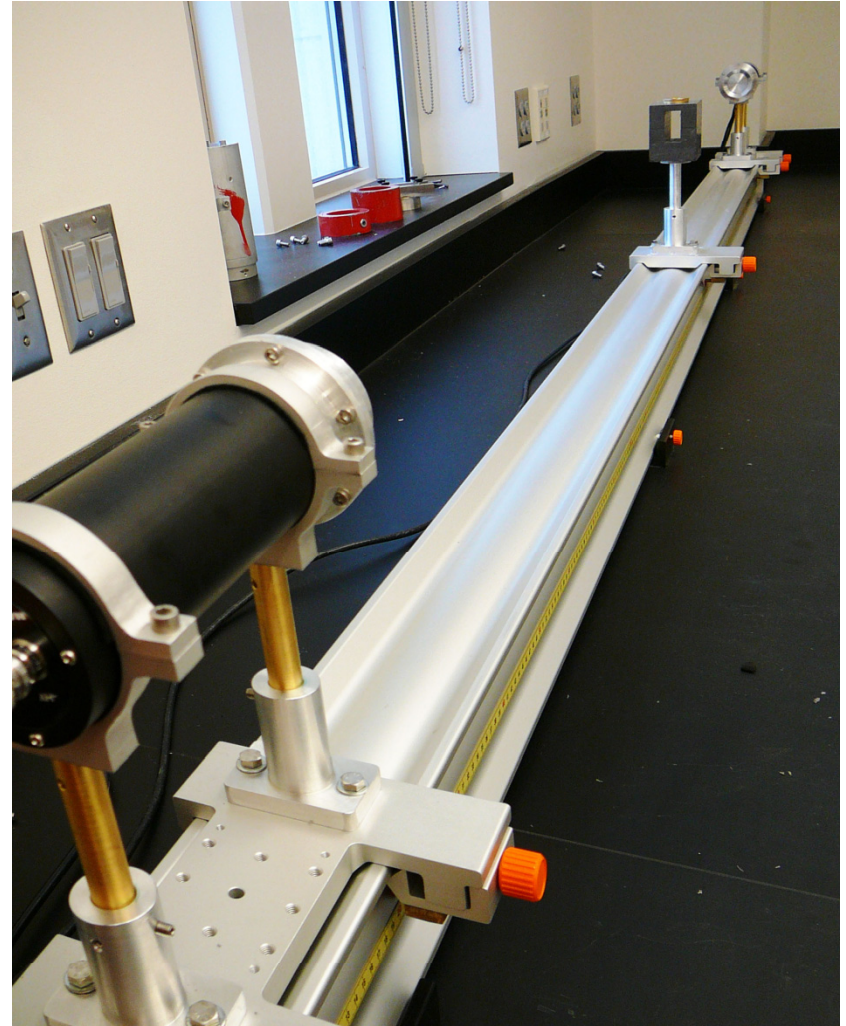
Experiment II XRF

- X Ray Fluorescence – Use an X-Ray source to excite atoms in the target instead of the proton beam.
- Measure a variety of objects.
- Test Moseley's Law relating X-ray Energy to atomic number.



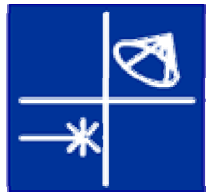
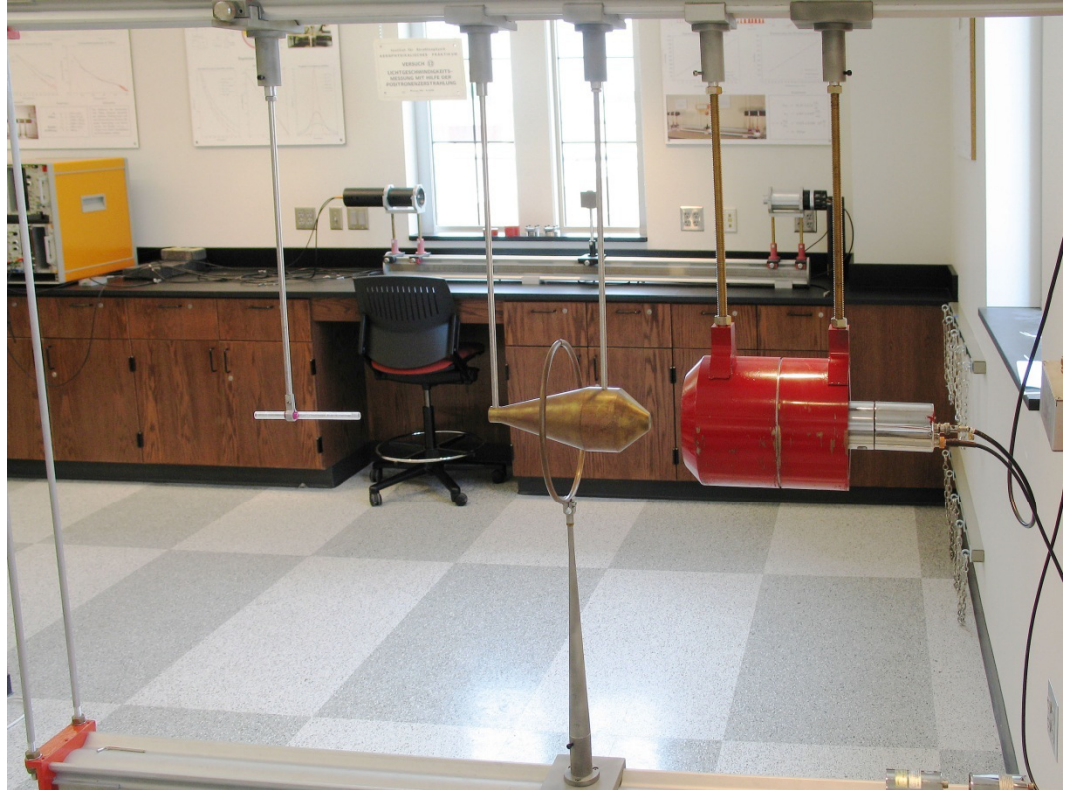
Experiment III Speed of Light

- Use Electronic modules as the stopwatch to time how long it takes a 511 keV photon to travel a distance of a meter or so.
- Use this information and a simple analysis to measure directly the speed of light.
- Photon source is the positron-electron annihilation from ^{22}Na .



Experiment IV Compton Effect

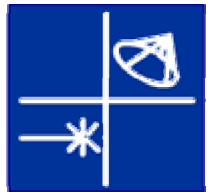
- An example of the particle nature of light by measuring the change in energy of photons scattered by the electrons of a solid object through different angles.



J I N A

Experiment V Optics

- A set of various labs with an emphasis on optical light.
- Look at wave nature of light through diffraction and interference.
- Also, see the effect of magnetic and electric fields on the motion of an electron
- Determine the electron's charge to mass ratio



J I N A